

Reaction of Keten Dithio-acetals with Aziridine. Iodide Ion-catalysed Rearrangement of Thio-aziridino-methylene Compounds to Novel Δ^1 -2-Methylthio-3,3-disubstituted Pyrrolines

By ARVIND KUMAR, HIRIYAKKANAVAR ILA and HIRIYAKKANAVAR JUNJAPPA*†

(Division of Medicinal Chemistry, Central Drug Research Institute, Lucknow, India)

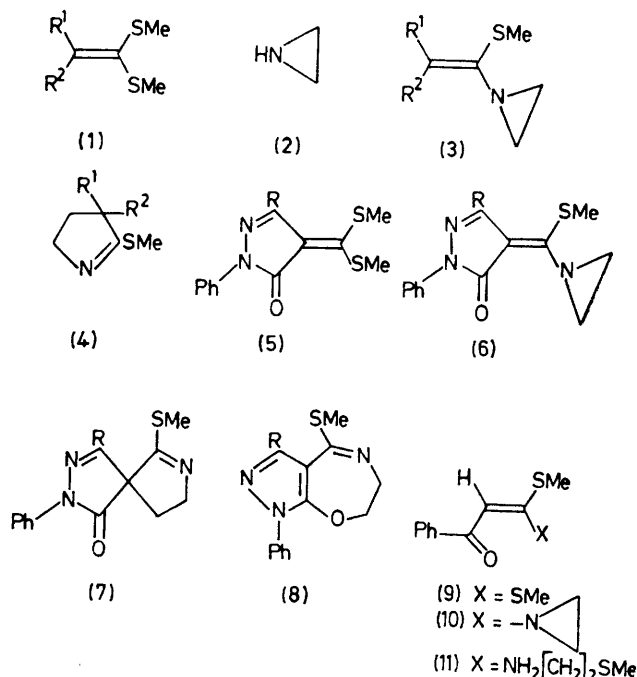
and SUSHILA MHATRE

(Department of Chemistry, Indian Institute of Technology, Bombay, India)

Summary The keten dithioacetals (1) and (5) react readily with aziridine (2) to give the corresponding thio-aziridinomethylene compounds (3) and (6) respectively, which on iodide ion-catalysed rearrangement, undergo ring expansion to give the pyrrolines (4) and (7) respectively.

THE isomerisation of aziridine derivatives has provided useful synthetic routes to a variety of heterocyclic ring systems.¹ However, *N*-vinylaziridines have not been studied extensively,² apparently because of their labile nature and the difficulty in their preparation. Although the polarised keten dithioacetals (1) are known to undergo ready displacement reactions with amines to give the corresponding thio-amino-methylene compounds,³ a similar reaction of these intermediates with (2) has not been reported. In continuation of our interest⁴ in the synthetic utility of keten dithioacetals, we report our preliminary results on the reaction of (1), (5), and (9) with (2) and further isomerisation of the intermediates (3) and (6) in the presence of iodide ion.

In a general procedure for the preparation of (3) and (6), a mixture of (1) or (5) (10 mmol) and (2) (50 mmol) in ether (150 ml) was stirred at 5–20 °C for 2–3 h. Removal of the solvent under reduced pressure yielded (3) and (6), respectively (Table). The n.m.r. spectra (CDCl₃) of (3a–d) and (6a–f) exhibited signals in the range δ 2.48–2.71



TABLE

	Compound (3) ^a				Compound (4) ^b	
	R ¹	R ²	M.p./°C	Yield/%	M.p. (b.p. at 1 mmHg)/°C	Yield ^c /%
a	CN	CO ₂ Et	96	87	(155)	85
b	CN	CONH ₂	105–106	89	125–126	80
c	MeCO	CO ₂ Et	Oil	70 ^d	e	e
d	CN	CN	109	85	f	f

	Compound (6) ^a		Compound (7) ^b	
	R	M.p./°C	M.p./°C	Yield ^e /%
a	Ph	111–113	144	80
b	<i>p</i> -MeC ₆ H ₄	115	145–146	80
c	<i>p</i> -MeOC ₆ H ₄	122–123	130–131	74
d	<i>p</i> -ClC ₆ H ₄	134	139–140	75
e	<i>p</i> -BrC ₆ H ₄	145–146	159–160	78
f	Me	119–120	80–82	78

^a Compounds (3) and (6) were characterized by i.r., n.m.r., and mass spectral data, and were analysed only for nitrogen, since they partially decomposed during purification. ^b Compounds (4) and (7) gave satisfactory elemental analyses. ^c Yields are for the I⁻ ion-catalysed reaction; thermal isomerization gave poor yields of products containing impurities. Compounds (7a–f) were isolated directly by diluting the reaction mixture with H₂O. ^d Isolated after stirring for 12 h. ^e Compound (4; R¹ = H, R² = CO₂Et), b.p. 105–108 °C at 1 mmHg, and not (4c), was isolated in 45% yield. ^f Polymeric products were obtained from the attempted isomerization of (3^d).

† Present address: Department of Chemistry, North Eastern Hill University, Shillong, Meghalaya India.

s, 3H, SMe) and δ 2.16—2.75br (s, 4H, aziridine). In a similar reaction of (9) and (2), unchanged (9) was obtained. However, when (2) and (9) were heated (165 °C; 24 h) in a sealed tube in benzene, (11)[‡] was formed exclusively (60%) instead of (10).

The I⁻ catalysed rearrangements of (3) and (6) were next examined. In a typical experiment, a mixture of (3a) (10 mmol) and KI (20 mmol) in dry acetone (150 ml) was stirred (20 °C; 2 h) and the reaction was monitored by t.l.c. The excess of solvent was removed, and the residue was diluted (H₂O) and extracted with CH₂Cl₂ to give (4a). Similarly (3b) yielded (4b), whereas (3c) underwent simultaneous deacetylation to give (4c) (Table). However, the attempted isomerisation of (3d) to (4d), both under thermal (refluxing C₆H₆, toluene, or xylene) as well as I⁻ catalysed conditions led only to polymeric products. Compounds

(6a—f), under both I⁻ catalysed and thermal conditions yielded the spiro-compounds (7) and no traces of (8) were detected (t.l.c., single spot).§¶

The formation of both (4) and (7) has not been observed in any of the rearrangements studied previously, involving aziridine ring expansion. The scope of this method is apparently governed by the ease with which (2) can react with keten dithioacetals to give the required *N*-vinylaziridines without the strained three-membered ring being destroyed.

The keten dithioacetals (1a—d)[§] and (5a—f)[§] were prepared by reported procedures.

One of us (A.K.) thanks C.S.I.R., New Delhi, for a junior research fellowship.

(Received, 26th April 1976; Com. 465.)

‡ Oil, separated by column chromatography; satisfactory elemental analysis; n.m.r. (CDCl₃): δ 2.15 (s, 3H, SMe), 2.45 (s, 3H, SMe), 2.73 (t, 2H, CH₂SMe), 3.61 (m, 2H, NHCH₂), 5.70 (s, 1H, vinylic), 7.43 (m, 5H, ArH), and 12.06br (s, 1H, NH).

§ Compounds (7a—f) showed strong absorption in the range ν_{\max} (KBr) 1690—1700 cm⁻¹ (C=O), which precludes the formation of (8).

¶ The n.m.r. spectra (Varian A60, CDCl₃), of (4a—c) and (7a—f) were fully in accord with their structures; e.g. (4a) showed signals at δ 1.35 (t, 3H, OCH₂Me), 2.50 (s, 3H, SMe), 2.80 (m, 2H, CH₂), and 4.28 (m, 4H, overlapping =NCH₂ and OCH₂Me), and (7a) at δ 2.52 (s, 3H, SMe), 2.74 (m, 2H, CH₂), 4.45 (m, 2H, =NCH₂), and 7.52—8.12 (m, 10H, ArH). Compounds (4b,c) and (7b—f) gave analogous data.

¹ H. W. Heine, *Angew. Chem. Internat. Edn.*, 1962, **1**, 528; R. S. Atkinson and C. W. Rees, *Chem. Comm.*, 1967, 1232; E. L. Stogryn and S. J. Brois, *J. Amer. Chem. Soc.*, 1967, **89**, 605; P. Scheiner, *J. Org. Chem.*, 1967, **32**, 2628; P. G. Mente, H. W. Heine, and G. R. Scharoubim, *ibid.*, 1968, **33**, 4547; A. Mishra, S. N. Rice, and W. Lwowski, *ibid.*, p. 481; P. G. Mente and H. W. Heine, *ibid.*, 1971, **36**, 3076.

² H. W. Whitlock and G. L. Smith, *Tetrahedron Letters*, 1965, 1389; *J. Amer. Chem. Soc.*, 1967, **89**, 3600.

³ D. Borrmann, in Houben-Weyl, 'Methoden der Organischen Chemie,' Band VII/4, 1968, Theime Verlag, Stuttgart, 1968, p. 421.

⁴ S. M. S. Chauhan and H. Junjappa, *Tetrahedron*, in the press.

⁵ A. Kumar, H. Ila, and H. Junjappa, *Synthesis*, 1976, 324.